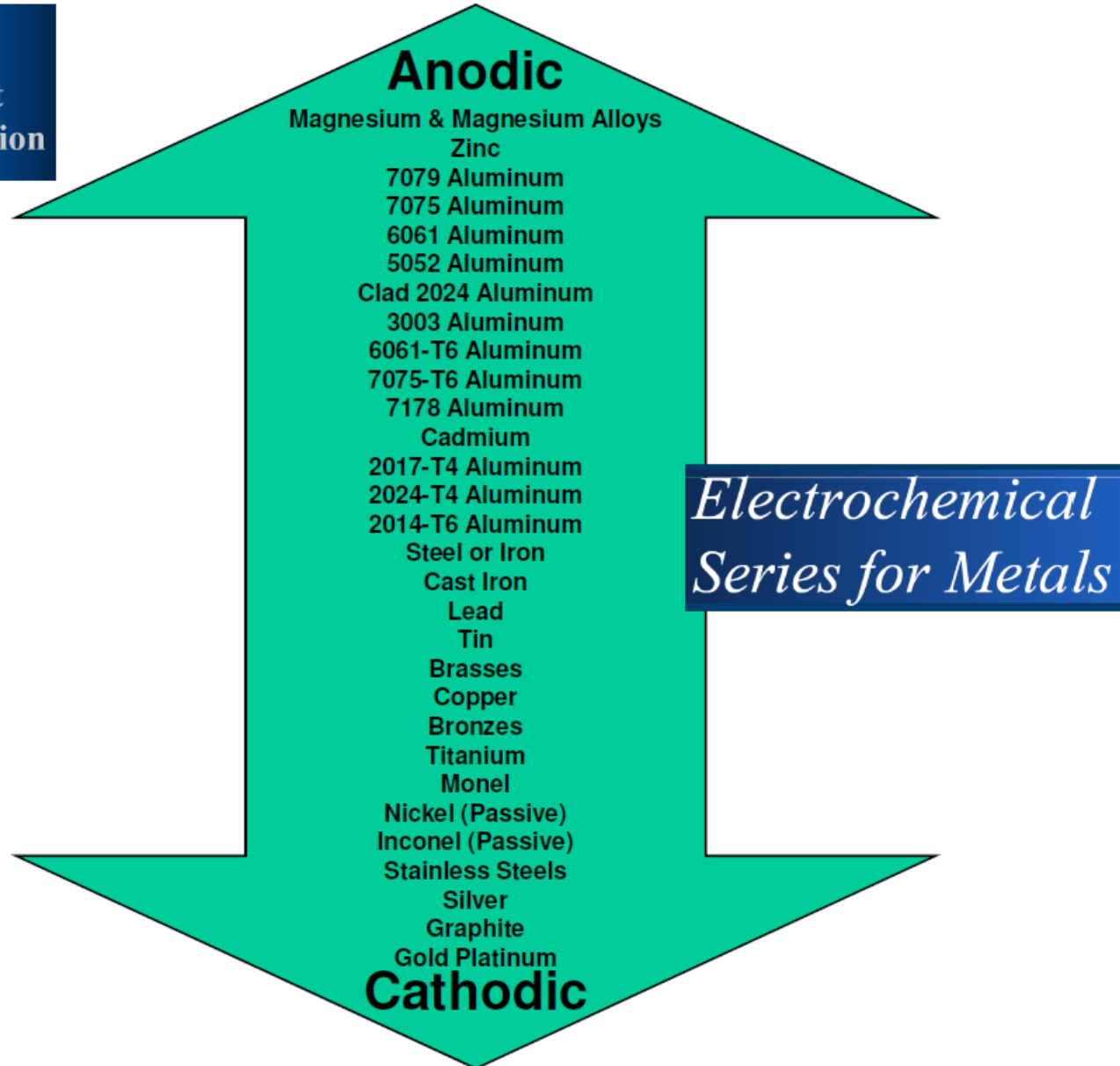


Priority #1



Protection against galvanic corrosion



The Galvanic Circuit

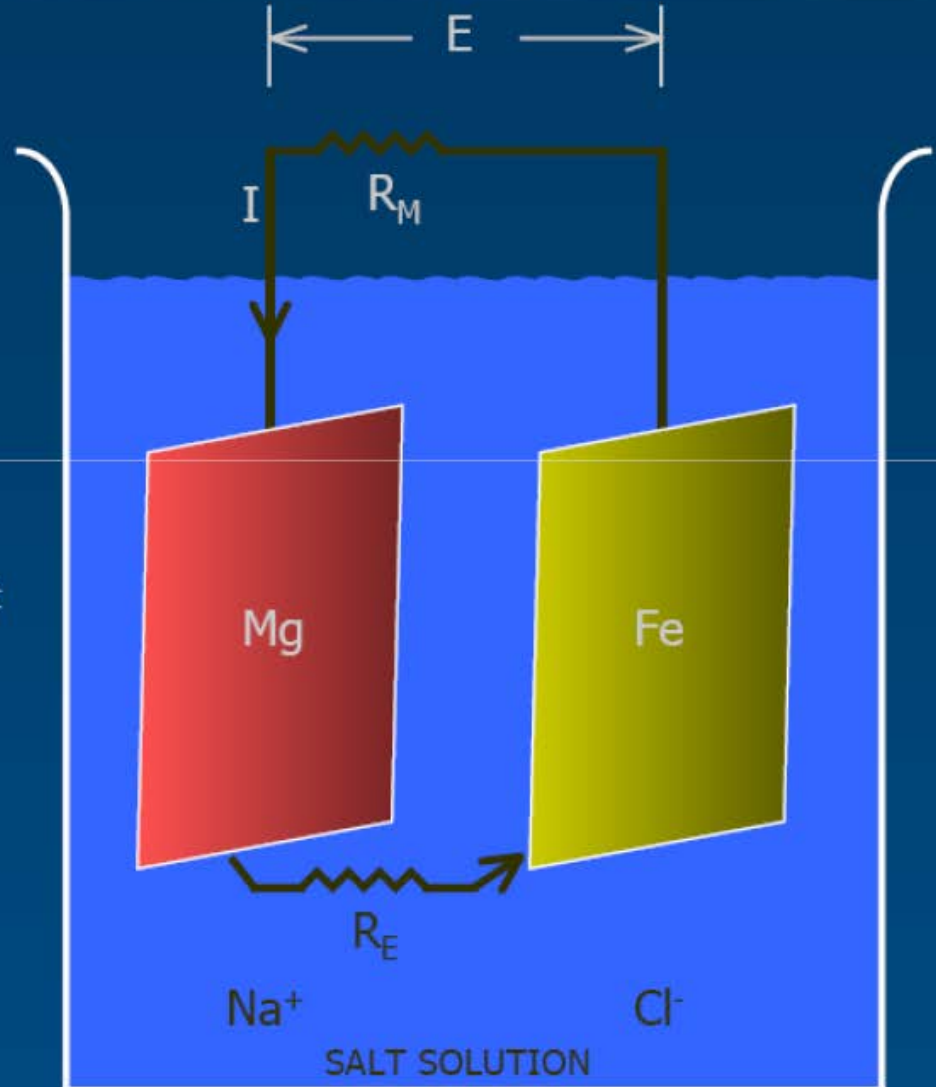
$$I = \frac{E}{R_E + R_M} \quad \text{where}$$

I = galvanic current

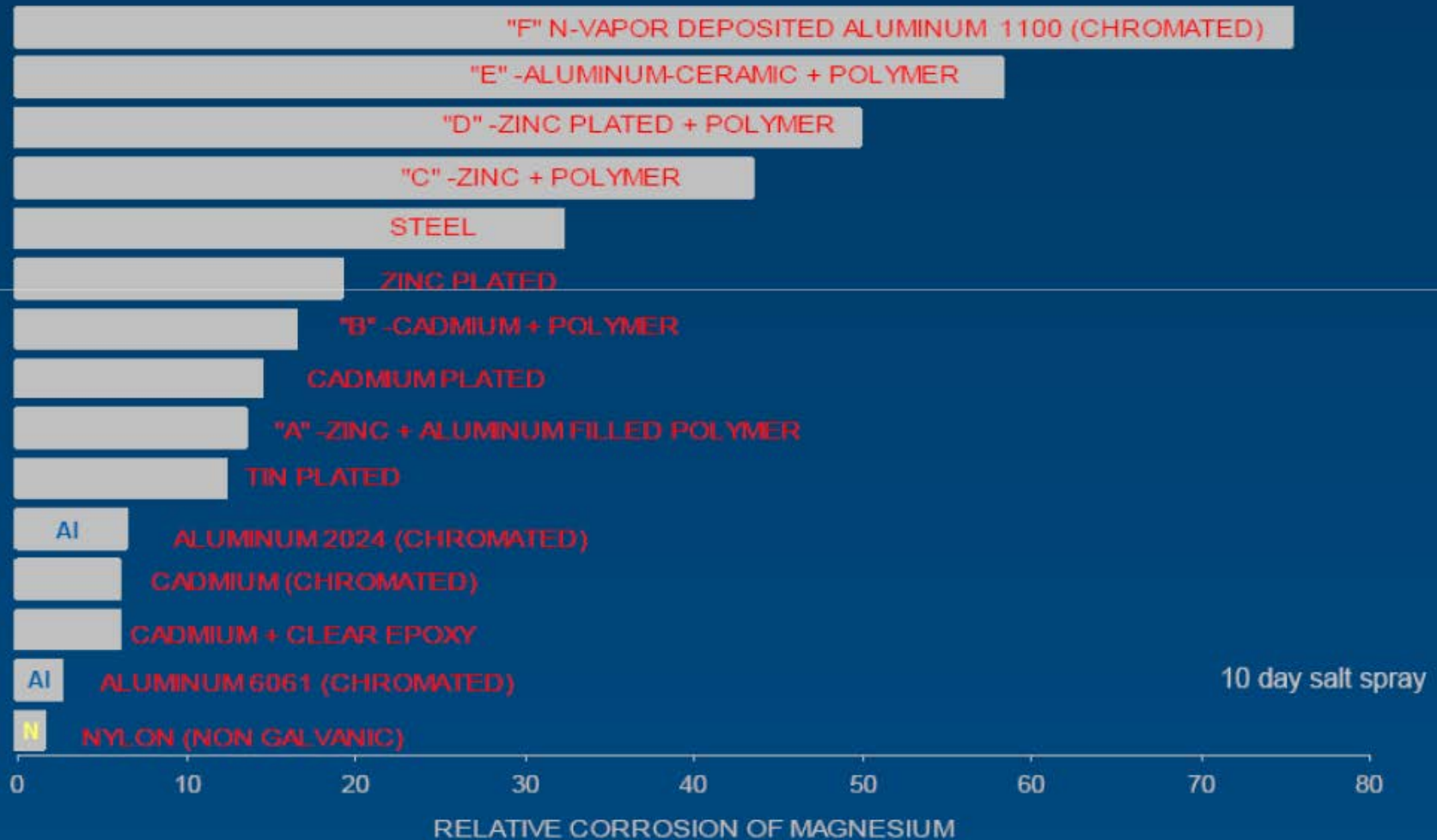
E = effective (polarized) difference between the magnesium and the dissimilar metal

R_E = resistance of the electrolytic part of the circuit

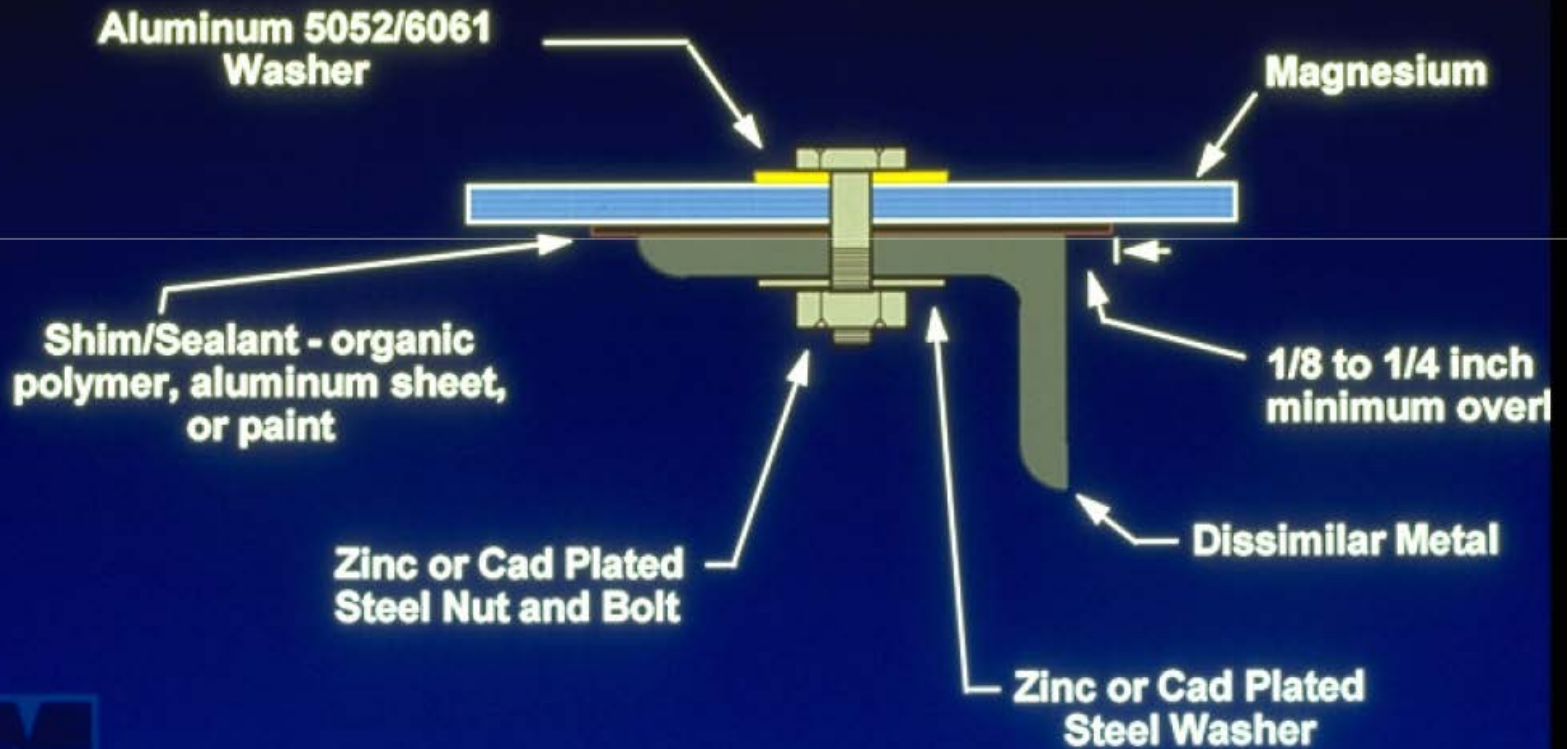
R_M = resistance of the metallic part of the circuit



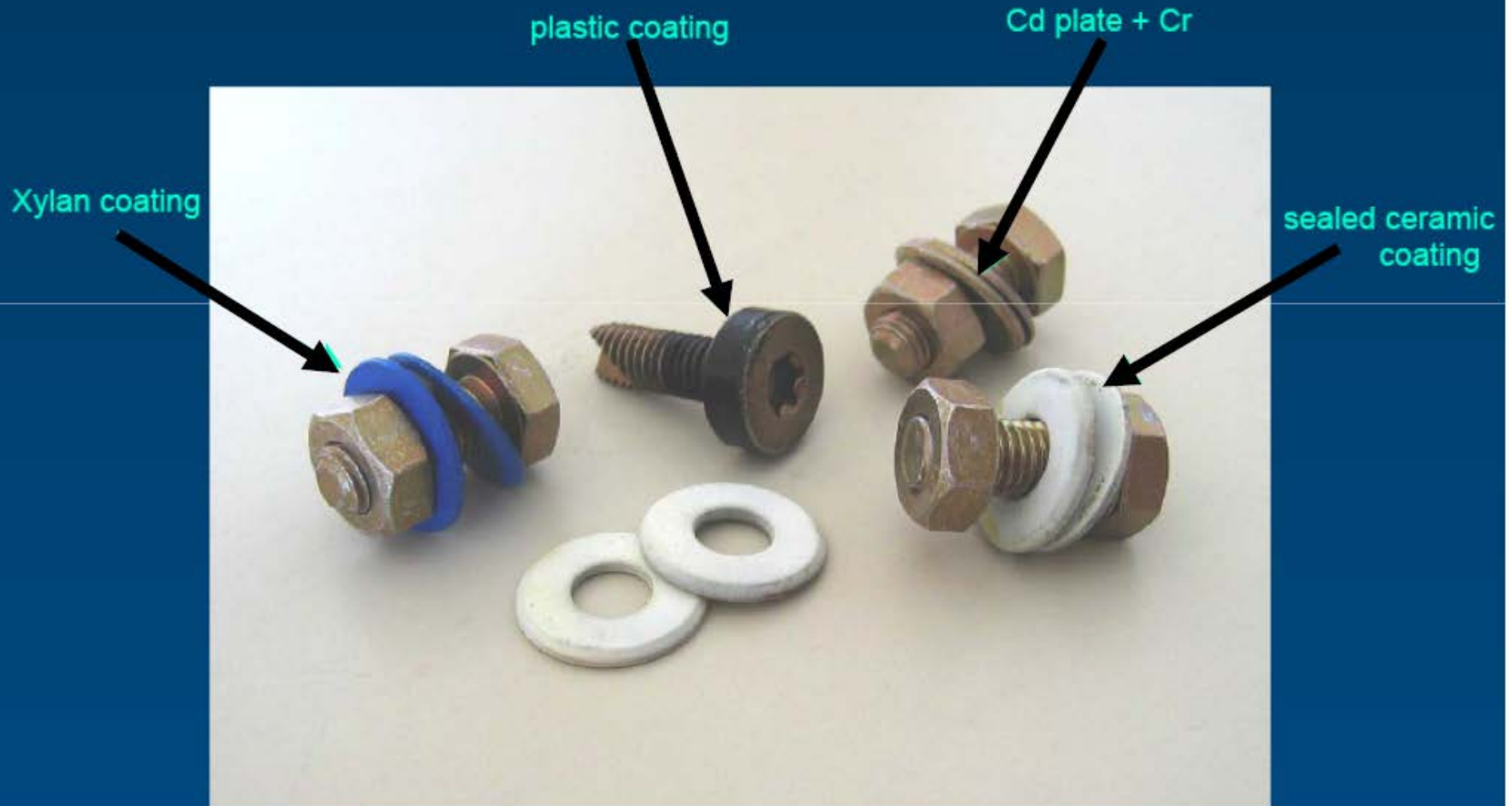
Galvanic Corrosion of AZ91D with Dissimilar Fasteners



PROTECTIVE MEASURES FOR BOLTED COMPONENTS



Fastener Choices

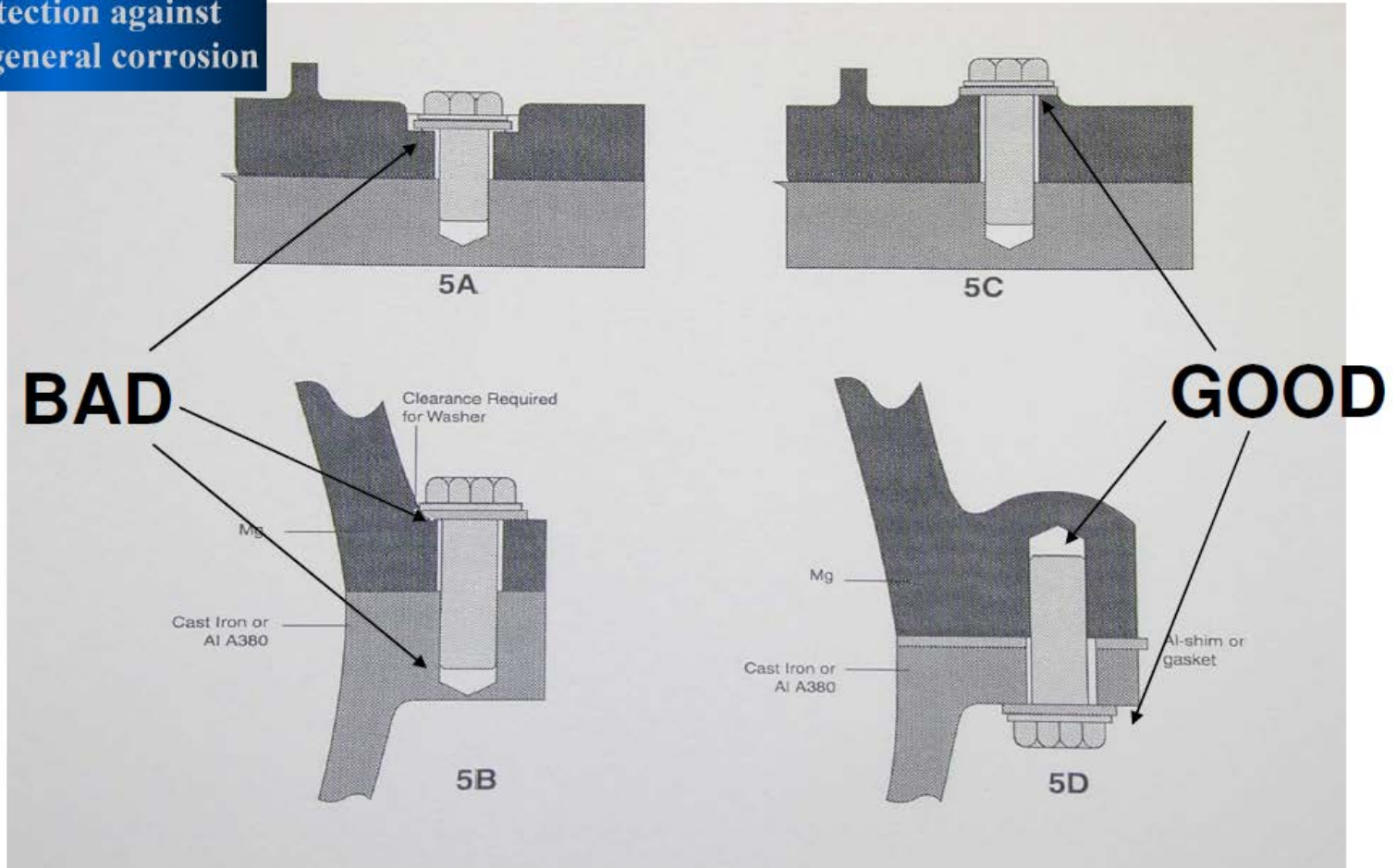


Priority #2



Protection against general corrosion

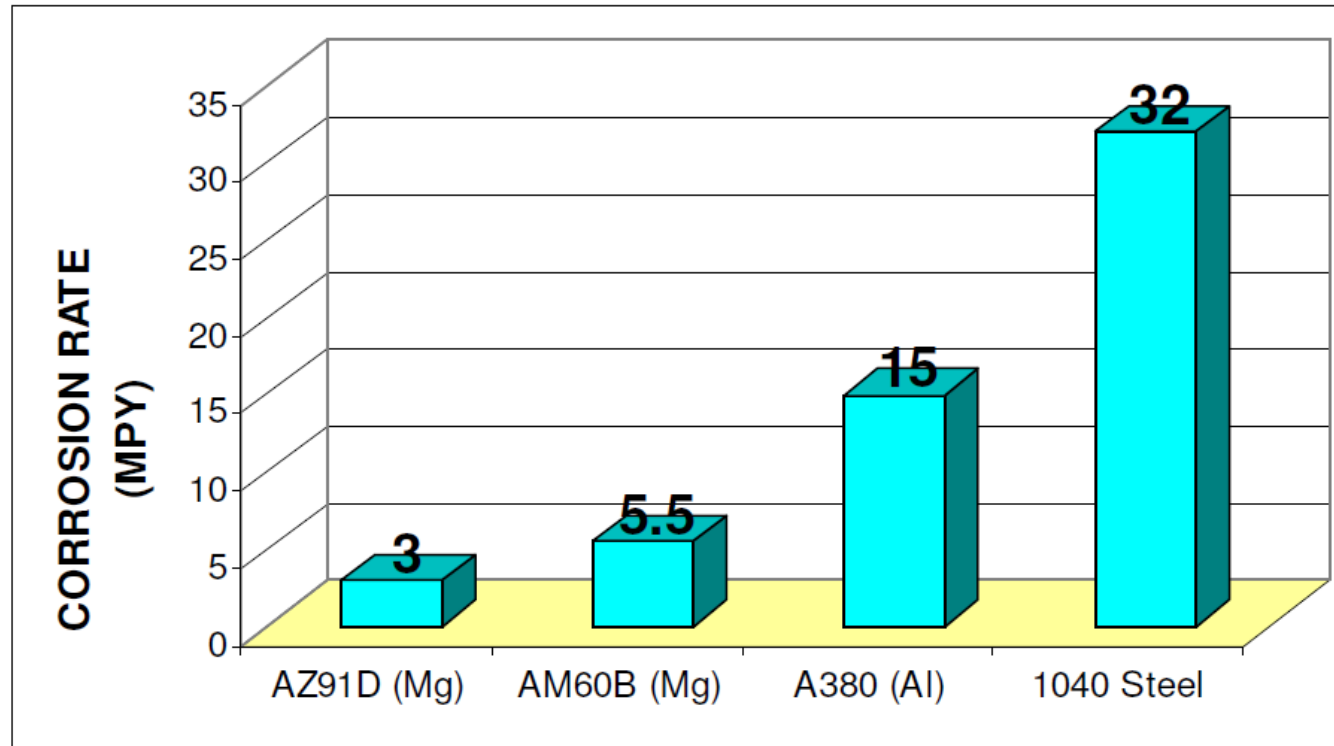
GOOD AND BAD DESIGNS



ELIMINATE ELECTROLYTE OR DIRECT CONTACT

* 10-DAY ASTM B-117 (Salt Fog) Test

SALT SPRAY CORROSION



Mg TXM Salt Spray Corrosion performance excellent

- *AZ91D : 3-4 Mils Per Year(MPY) lost in ASTM B-117 Test Panels*
- *4 - 5 X better than Cast Aluminum Alloys*
- *8-10 X better than Steels__*

SALT SPRAY TEST PANELS

Corrosion after 10-day Salt spray test



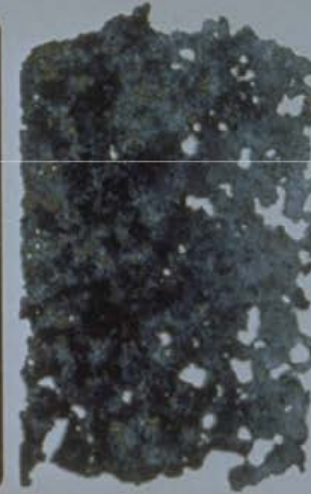
AZ91D
4mpy



380AL
13mpy

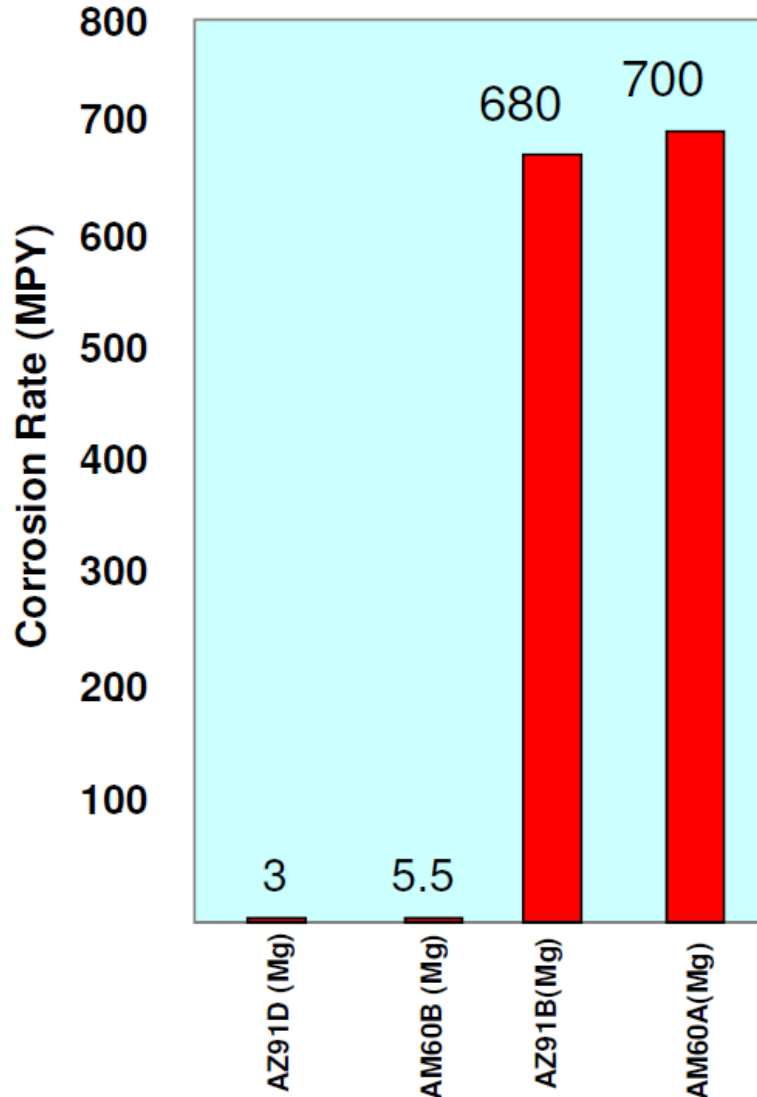


**Carbon
Steel**
30mpy



AZ91B
680mpy

ASTM B-117 Results



HIGH PURITY ALLOY BENEFIT

- Corrosion “issues” of 1960’s vintage alloys no longer dominant to designs
- High Purity Alloys (AZ91D, AM60B) perform 100 - 200x better than “old” alloys (AZ91B, AM60A)
- Why ?
 - Alloy chemistries modified
 - “Tramp” Element Content reduced
 - Iron (Fe)
 - Copper (Cu)
 - Nickel (Ni)
 - Internal Galvanic Effects Eliminated